## ANALYSIS OF FOOD PRODUCTS

FNH 4114/6114 **FALL 2017** 

**Instructor:** Dr. Wes Schilling

Office: Room 106 Herzer Building

Phone: 662-325-9456

e-mail: schilling@foodscience.msstate.edu

Class Time: Monday and Wednesday 11:00 am- 11:50 am

**Laboratory Time:** Monday 2:00-4:50 pm;

Yan Campbell, Sensory Laboratory 112, yl479@msstate.edu, phone (325-8366) **Laboratory Coordinator:** 

**Teaching Assistants/:** 

Morgan Von Staden, Herzer 156, mev63@msstate.edu **Guest Lecturers** 

**Office Hours:** Available anytime; you should feel free to call prior to making a visit to make

sure that we are in our offices and available. Questions may also be submitted

by e-mail.

Textbook: Food Analysis, fifth edition.

http://www.barnesandnoble.com/s/Food-

Analysis?keyword=Food+Analysis&store=textbook, also available at

amazon.com

A bound laboratory notebook with numbered pages and duplicate pages for

carbon copies is required.

Calculator (regression, statistics) is needed for all laboratories and exams.

Several handouts will be provided.

Software: Students will be required to use spreadsheets for data handling and graphical

> presentations. Microsoft Excel is preferred but students may use any spreadsheet that has regression analysis subroutines and graphical output.

Attendance: Attendance is mandatory for all laboratory sessions. Excused absences will only

be granted by the instructor <u>prior</u> to the start of the laboratory session.

Unexcused absences will result in a "zero" grade for all assignments associated

with that lab.

**Objective:** This course is designed to acquaint students with the basic laboratory and

problem solving skills used in modern food analysis by a combination of

interactive **LECTURES** and hands on laboratory experience.

I support the Honor Code at Mississippi State University and expect all students **Honor Code:** 

to work independently and abide by all its policies.

Lab. Protocol: Be on time to lab.

Do not wear open toe shoes to lab.

A lab coat is recommended, but not required.

Read the lab in advance and include all calculations, tables, and flow diagrams

in your lab notebook as required prior to starting lab.

# **LABORATORY SCHEDULE (Subject to Change)**

<b>DATE</b>		TOPIC (Currently)
AUGUST	21	Intro lab and Data Analysis
	28	Lab 2. Moisture, Water Activity and Proximate Analysis.
SEPTEMBER	4	Holiday (No lab).
	11	Lab 3. pH Determination / Buffers.
	18	Lab 4. Spectroscopy: Visible Spectra.
OCTOBER	25	Lab 5. Spectroscopy: Extinction Coefficient & Standard Curve.
	2	Fall Break Week (No lab scheduled)
	9	Lab 6. Colorimetric Protein Assays.
	16	Lab 7. Enzyme as Reagents.
	23	Lab 8. Texture Analysis (Instron).
NOVEMBER	30	Lab 9. Sensory Evaluation.
	6	Lab 10. Gas Chromatography.
	13	Lab 11. Sensory Evaluation part 2
	20	Lab 12. Industrial Applications

### LABORATORY REPORTS

Written laboratory reports are required for this course. All laboratory reports are due at the beginning of your next laboratory session. Any laboratory report submitted by 11 am on the Wednesday prior to the next lab will be considered early and receive a 10% bonus. **Reports not turned in at start of lab will be considered late, and late lab reports will not be accepted.** Exceptions to the above policies will only be granted when arrangements have been made <u>prior</u> to the original Monday deadline. Normally, exceptions will be granted only in those situations accepted for missed exams, below.

Do not share returned, graded lab reports or computer discs with files containing your laboratory report with fellow students. If a fellow student copies your report, the Honor Court will also charge you with a violation.

## **GRADES**

All missed examinations will be given a grade of "zero" with the following exceptions: (a) where arrangements have been made <u>prior</u> to the examination; (b) illness (doctor's excuse required); © serious family problems or illness (documentation required). Your final grade will be based upon the following scale:

Lecture Examination #1		15%
Lecture Examination #2		15%
Final Examination		20%
Laboratory Reports (8)		30%
Homework Assignments		10 %
Attendance		5 %
Participation		5 %
		-
	TOTAL =	100%

**NOTE**: THE FINAL EXAMINATION WILL BE COMPREHENSIVE.

# TENTATIVE LECTURE SCHEDULE

August 16 Welcome and Introduction to Food Analysis 21, 23, 28 Evaluation of Analytical Data (Standard curve Information)	
21, 23, 28 Evaluation of Analytical Data (Standard curve Information)	
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Evaluation of Analytical Data/Sampling and Sample Preparation, Solvir	ıg
Problems, Units Review	
30, Sept 6 Moisture Analysis/Proximate Analysis	
September	
8 pH and Titratable Acidity, Introduction to Flavor	
pH and Titratable Acidity, Introduction to Flavor; Lab 3 (Principles),	
lecture Spectroscopy UV-VIS Spectroscopy and Infrared Spectroscopy	
Test 1 Review, Lab 5 Spectroscopy lecture	
22 Test 1	
27 Atomic Absorption Spectroscopy, Mineral Analysis, Test Review!	
October 2 Other Protein Analyses	
4 Lecture 6 lab, Lipid and Carbohydrate Analysis	
9 Enzyme Kinetics, Lab 7 Lecture	
11 Principles of Chromatography	
16 Gas Chromatography/ Flavor Analysis	
High Performance Liquid Chromatography, Lab 9 Lecture	
23 Texture Analysis, Lab 10 Lecture	
25 Test 2 review	
30 Test 2	
November 1 Sensory Evaluation, Lab 10 Lecture	
6 Sensory Analysis, Color Analysis	
8 Color Analysis, Analysis of Dairy Products, (May make beef jerky, if we	have
time and interest)	
13 Analysis of Meat Products	
15 Industry Examples and Stories	
20 Industry Examples and Stories	
November 27 Review for Final (Possibly take home)	

## Data Analysis (Things that you must be able to do)

- 1) determine the mean (by hand and using excel)
- 2) determine the standard deviation (by hand and using excel)
- 3) Know why you run replicates
- 4) Have an appreciation for the importance of sampling correctly and choosing the correct technique
- 5) determine the relative standard deviation (by hand and using excel)
- 6) determine the standard error of the mean (by hand and using excel)
- 7) determine confidence intervals for both the normal and t-distributions (by hand and using excel)
- 8) Understand your sources of error
- 9) Know the definitions of specificity, and sensitivity
  - a. What determines sensitivity in the standard curve?
- 10) Figures of Merit for the Standard Curve
  - a. Limit of Detection (Determine using Excel)
  - b. Linear Range (Determine using Excel)
  - c. Sensitivity (Determine using Excel)
  - d. R-square (Determine using Excel)
- 11) Determine the equation of line for a standard curve
- 12) Determine values for unknown samples using a standard curve
- 13) Graph standard curves using Excel
- 14) Use appropriate figure captions on your standard curve graphs
- 15) Know how to appropriately report data with significant digits
- 16) Know when to reject data (Q stat)

```
x_i = observation i

i=1,2,...n

n=total number of observations

y = response (absorbance, peak area)

m=sensitivity

x=concentration

b=y-intercept
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Standard curve information

## Learning Outcomes and Objectives for Food Analysis

# There are three parts to Food Analysis: sample preparation, instrumental analysis, and data analysis

## Food Analysis Lecture:

- 1) You should be familiar with the names of the instruments and methods that are used to determine the composition and chemical properties of foods
- 2) You should able to understand the general chemistry behind how the instruments work, but you should by no means be an expert
- 3) You should have an understanding and appreciation of the importance of sample preparation in food analysis for methods commonly used in food analysis
  - a. Homogenization
  - b. Storage
  - c. Preparation of samples for UV-VIS spectroscopy, Infrared spectroscopy, atomic absorbtion spectroscopy pH, moisture analysis, chromatography (gas, liquid, paper, thin layer)
- 4) Understand how the applications mentioned in 3c are utilized in the food industry
- 5) Be able to take data that you receive from an analysis and know what to do with it
  - a. Is it right
  - b. Did you take enough samples
  - c. Determine the appropriate mean (concentration)
  - d. Determine the appropriate standard deviation (concentration)
  - e. Determine the appropriate relative standard deviation (concentration)
  - f. Run regression on standard curves
  - g. Determine unknowns
  - h. Perform calculations (chemistry)
- 6) Improve your ability to solve problems as well as improve your ability to understand what the problem is?
- 7) You should know exactly what infrared spectroscopy is used for in the food industry (you should also never forget this)
- 8) You should be able to locate the correct method when asked to make a determination about a food product

At the end of this course, you should be able to do the following in the laboratory:

- 1) Record the correct information in a laboratory notebook in an appropriate manner (including visual observations, etc.)
- 2) Perform dilution tables
- 3) Read and understand laboratory methods prior to attending laboratory classes
- 4) Improve your ability to work on a laboratory benchtop
- 5) Understand the objectives of your laboratories
- 6) Understand how to handle (what to do with) data in your laboratories
- 7) Understand the basic statistics necessary in Food Analysis. To be able to calculate these statistics if you have access to the data and the formulas. To be able to locate the correct formulas
- 8) Understand the importance of sample preparation
- 9) Understand the purpose of using different equipment and the basic theory behind how they work. Also understand how to tell if data is accurate and what to do with the numbers that the equipment provides to you.

#### Class rules

- 1) You should not talk among yourself during lecture time unless you are working in groups. (1<sup>st</sup> offense, you will be asked to stop, 2<sup>nd</sup> offense, you will be asked to leave)
- 2) You should show up to the laboratory class on time.
- 3) You should not miss class, especially labs without an appropriate excuse that is cleared with the instructor prior to class.
- 4) You should not use your cell phone during lecture time. You can use your cell phone in lab to time experiments and when you are at a desk during waiting time in labs. However cell phones, should not be used on lab benches and when someone is teaching in front of the chalkboard during lab time.

## APPLIED FOOD CHEMISTRY

# <u>FNH 4241/6241</u> <u>SPRING 2018</u>

**INSTRUCTORS:** 

Dr. Wes Schilling

Office: Herzer Building, Room 106

Phone: 325-9456

Email: Schilling@fsnhp.msstate.edu

**OFFICE HOURS:** 

Based on appointment, you should feel free to call or email prior to

visiting to make sure that I am in my office and available.

LABORATORY COORDINATOR:

Dr. Yan Campbell

Office: Garrison Sensory Evaluation Lab Room 112

Phone: (662) 325-8366

Email: ycampbell@foodscience.msstate.edu

**TEACHING ASSISTANT:** 

Morgan Von Staden Office: Herzer Room 156 Phone: (662) 325-7698 Email: mey63@msstate.edu

**TEXTBOOK:** Hardbound Laboratory Notebook (with tear-out carbon pages)

Software for Word Processing and Spreadsheet (For example: Microsoft Word and Excel)
Three ring binder for returned lab reports

**HONOR CODE:** I do support and abide by the Mississippi State University Honor

Code. In this laboratory you will work with a group of fellow students to complete each week's assignment. In many cases, you may also share group data among individual students. However, each assigned laboratory report is graded and each student is expected to work independently in performing calculations, making observations and in preparing the final written report. Finally, protect yourself and do not allow fellow students to

borrow disks containing your lab reports.

#### LABORATORY GUIDELINES

- 1. You will be provided with a laboratory manual at the beginning of the semester. Be sure to read the laboratory protocol for each week's lab. Prior to class time so that you understand how to complete the experiment in a timely manner.
- 2. Proper preparation for each laboratory means that you have read and understood the laboratory handout prior to coming to class. Prior preparation will allow you to spend considerably less time in the laboratory actually performing the experiment.
- 3. Prior to beginning work on an experiment, gather together all of the equipment and materials that you will need to carry the laboratory exercise to completion. The lack of equipment or reagents at a critical time may lead to poor results or failure of the experiment. Practice good laboratory management principles.
- 4. All students are encouraged to wear laboratory coats or smocks to protect clothing as well as protective eyeglasses. Arrangements can be made to store your eyeglasses in the laboratory, if you wish. Please contact your laboratory instructor. Appropriate clothing and shoes (low heels and closed toes) must be worn in the lab. No smoking, eating or drinking is allowed in the lab at any time (except during sensory evaluation). Keep cell phones in the silent mode (or turned off).
- 5. At the conclusion of your experiment, please rinse all dirty glassware and place in the appropriate place; make sure your laboratory bench is left clean and in order.

#### **GRADES:**

TOTAL POINTS	2000
Quizzes	300
Participation	100
Laboratory Notebook	400
Laboratory Reports	1200

Grades: A=1800-2000 points

B=1600-1799 points C=1400-1599 points D=1200-1399 points F= <1200 points

#### LABORATORY REPORTS:

- 1. All assigned laboratory reports can be submitted electronically by e-mail or as a printed copy by the beginning of class time. **Laboratory reports not submitted promptly by class time will be considered late.** Ten percent (10%) of the total possible grade will be deducted for each class day (24 hrs) that the report is late to a maximum of 50%.
- 2. The entire report (including all text and tables) must be prepared using word-processing software (for example, Microsoft Word). All graphs must be generated using computer spreadsheet programs (for example, Microsoft Excel) and incorporated into the word-processing file. Reports, which do not look professional, will be returned; a late penalty will be imposed until a satisfactory laboratory report is submitted.

#### LABORATORY NOTEBOOKS:

- 1. The purpose of a laboratory notebook is to accurately and legibly record experimental data and observations. Laboratory results are only as valuable as their written record. You should start now to develop good techniques recording in your laboratory notebook. All data obtained during laboratory must be recorded in a hardbound notebook. Copying from pieces of paper must be avoided because of possible copying errors, wasted time, and the danger of losing papers containing data. A carbon copy of all your experimental data must be submitted at the conclusion of each lab period.
- 2. The grade for your laboratory notebook will be based upon the following criteria:

You arrived prepared for the laboratory session and brought your laboratory notebook and laboratory manual.

All entries must be made in your notebook using an ink pen; no pencil entries allowed. The first page should contain your name, the names of your laboratory partners, a title for the experiment and the date on which it was performed.

Tables for recording your data should be prepared in advance of the laboratory session. Allow space for descriptive terms and remarks. Reference all tables and observations to the specific step in the experimental procedures described in your laboratory handout.

Any step in the procedure that involves weighing or measuring reagents, products or samples should be referenced to the step in the experimental procedures described in your laboratory handout. Be sure to indicate amounts or volumes of samples or reagents used in conducting your experiment. Be sure to also record any deviations from the experimental protocol described in your handout.

Be sure to record any changes in sensory characteristics of a product that occur while conduction your laboratory experiment. Again, be sure to reference all observations to steps in the laboratory procedures.

# LABORATORY SCHEDULE: Mondays 3:00 - 4:50 pm

<u>DATE</u>	EXPERIMENT

January 22th Lab 1. Major Food Components in Milk

January 29th Lab 2. Non-Enzymatic Browning

February 5th Lab 3. Carbohydrate Crystallization

February 12th Lab 4. Starch/Gelatinization & Gelation

February 19th Lab 5. Pectin Gels

February 26<sup>th</sup> Lab 6. Protein Characterization / Properties

March 5<sup>th</sup> Lab 7. Soy Protein Functionality

March 12<sup>th</sup> -16th SPRING BREAKூ

March 19th Lab 8. Flour Protein/Dough Formation

March 26th Lab 9. Lipid Absorption

April 2th Lab 10. Lipids/Shortening Ability

April 9th Lab 11. Chromatography

April 16th Lab 12. Candy Production

## FNH 8163 FLAVOR AND FOOD ACCEPTANCE SPRING 2015

Instructor: Dr. Wes Schilling (Lecture)

Office: Room 104 Herzer Building

Phone: 325-2666

e-mail: schilling@foodscience.msstate.edu

Class Time: Lecture Classes: 11:00-12:15 pm Monday, Wednesday

**Office Hours:** Open office hours; you should feel free to call prior to making a visit to make

sure that we are in our offices and available. Questions may also be submitted by

e-mail.

Textbook: Meilgaard et al. 1991 (4<sup>th</sup> edition) or 1999 (4<sup>th</sup> edition): Sensory Evaluation

**Techniques** 

Suggested Readings from Flavor Applications (Source Book of Flavors,

1994, 1999)

Several handouts and readings will be provided

**Software:** Students will be required to use spreadsheets for data handling and graphical

presentations. Microsoft Excel is preferred but students may use any spreadsheet that has regression analysis subroutines and graphical output. Access to and use of Statistical Analysis Software (SAS) will also be required. Examples of how to

analyze sensory data and chromatographic data will be covered.

**Attendance:** Attendance is mandatory for all sessions. Excused absences will only be granted

by the instructor prior to the start of the class. Unexcused absences will result in

a "zero" grade for all assignments associated with that class.

**Objective:** This course is designed to acquaint students with the basic understanding of food

and flavor acceptance characteristics from both a sensory evaluation standpoint and an instrumental standpoint as well as to relate the two of these together by a combination of interactive **LECTURES** and **HANDS ON** research experience. This class will cover data collection and statistical analysis for this type of work.

Learning Outcomes: For students to have a general understanding of sensory science,

instrumental analysis of flavor volatiles, and flavor chemistry. For students

to be able to understand how to conduct sensory evaluation tests,

instrumental volatile analysis, experimental design, statistical analysis, and interpretation of sensory and volatile analysis data. For students to be able to understand the principles that are used as well as to perform the basic statistical analyses that are used to analyze consumer sensory data, trained

sensory data, and gas chromatography data.

**Honor Code:** I support the Honor Code at Mississippi State University and expect all students

to work independently and abide by all its policies.

**CLASS PROJECTS** 

<u>Class Project Number 1</u>: You and 1 or 2 other students will pick a food product that you are interested in and you will determine the volatile compound profile, odor active compounds present, sensory descriptors, and consumer acceptability for that product. You will need to meet with Dr. Yoon and I to discuss the project. We will help you design the project. Then you will conduct the experiment and you will have a 3-5 page paper. Your paper will only consist of a Materials and Methods, data analysis section and graphs or tables representing your data, and students should work together to conduct the research and statistical analysis (as appropriate) but should work alone on their write-up of the project.

<u>Class Project Number 2:</u> You and 1 or 2 other students will receive a comprehensive data-set from research conducted at Mississippi State University. As a group, you will be required to conduct statistical analysis on this research. Then, each student should work on their own to write a results and discussion paper on the dataset, demonstrating that they have an understanding of the acceptability of the food products as well as a general understanding of the data.

#### HOMEWORK ASSIGNMENTS

- 1) Paper Review and Discussion: Each student will be provided with a research paper in which they will be required to conduct a written review of the paper. The student will review the paper as if they had received it from an editor (Journal of Food Science, Journal of Sensory Studies, etc.) and were asked to determine if it is acceptable for publication. The student will also give an oral presentation where they will discuss the paper for 10-15 minutes and describe the objectives of the study, how the study was conducted, and the major findings of the study.
- 2) Three short assignments will be given related to laboratory/hands-on classes that deal with thresholds, difference tests, descriptive tests, texture profile analysis, and GC analyses.

#### **GRADES**

All missed examinations will be given a grade of "zero" with the following exceptions: a. where arrangements have been made <u>prior</u> to the examination; b. illness (doctor's excuse required); c. serious family problems or illness (documentation required). Your final grade will be based upon the following scale:

Midterm Examination Final Examination Participation Attendance Homework Assignments Paper Review Written (5 %) Paper Discussion Oral (5 %) Lab Assignments (3) (10 %)		20% 20 % 5 % 5 % 20 %
Projects (2)		
Project 1		15 %
Project 2		15 %
Statistical Analysis (5 %)		
Written Paper (5 %)		
Oral Presentation (5 %)		
	TOTAL =	100%

**NOTE**: THE FINAL EXAMINATION WILL BE A COMPREHENIVE TAKE HOME EXAMINATION.

Course Schedule (Subject to Change)

Class Topic	Date	Reading Assignment
1) Intro to Class (Course Expectations, History	January 8th	Powerpoint, Syllabus
of Sensory Evaluation and Flavor Chemistry)	•	• •
2) Importance of Statistics in Sensory Science,	January 10th	Powerpoint, Handout
Flavor Chemistry, and Food Science	•	-
3) Sensory Attributes and Perception,	January 17th	Powerpoint, 8-22, Ch 2
Test Rooms, Product, and Panel	•	23-36, Ch 3
Factors Affecting Sensory Verdicts		37-42, Ch 4
4) Overall Difference Tests	January 22nd	Powerpoint, 60-98, Ch 6.1
5) Attribute Difference Testing	January 24th	Powerpoint, 99-122, Ch 6.2
6) Determining Thresholds (Lab Assignment)	January 29th	Powerpoint, 124-133, Ch. 7
7) Affective Tests/Consumer Testing	January 31st	PowerPoint, 201-226, Ch. 8
8) Consumer Testing/Cluster Analysis	February 5th	PowerPoint, Journal paper*
9) Selecting and Training Panel Members	February 7th	Powerpoint, 135-151, Ch. 8
10) Descriptive Analysis Techniques	February 12th	Powerpoint, 188-200, Ch. 9
11) Descriptive Analysis Techniques	February 14th	
Class Activity (Lab Assignment)	•	
12) Labeled Magnitude Scale and Ranked Data	February 14th	Powerpoint, 43-57, Ch. 5
Class Activity	•	1 , , ,
13) Midterm	February 19th	
14) Flavor Industry	February 21st	Handout 1 Source Book of Flavors (SBF)
15) Flavor Chemistry	February26th	Handouts 2 and 3 SBF
16) Volatile Extraction Techniques	February 28th	Handouts 2 and 3 SB1 Handouts 2 and 3
17) Instrumental Analysis of Volatiles	March 5th	Handouts 2-3 SBF
18) Merging Instrumental Analysis with Flavor	March 7th	Handouts 2-5 SBF, etc
Chemistry	Water 7th	Trandouts 2 3 SBT, etc
Spring Break	March 10-18th	
19) GC Lab-SPME, Purge and Trap and GCO	March 19th	Lab Handout, Lab report
20) How to Review a Research Paper	March 21st	Powerpoint
Class Discussion Project 1	Water 21st	Towerpoint
21) Principal Components Analysis	March 26th	Powerpoint, Handouts
21) Preference Mapping (Descriptive/Consumer)	March 28th	Powerpoint, Handouts
22) Real World Applications-Dry Cured Ham	April 2nd	PowerPoint PowerPoint
23) Real World Applications-Yogurt, Cheese	April 4th	Powerpoint
Class Project 2 Discussion	при чи	Towerpoint
24) Research Paper Reviews Due	April 9th	
25) April 12th-25th Assistance with Projects, By ap		cheduled meetings twice a
week		2
26) Final Oral Presentations	April 25th	
Receive Take Home Test (Due by noon on Ma		
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Final Exam: Discussion Question test showing that you have learned concepts discussed pertaining to sensory evaluation, experimental design, and flavor chemistry